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## Buoyancy Worksheet

1) A medical ship weighs $100,000 \mathrm{~kg}$. What volume of fresh water ( $\rho=1025 \mathrm{~kg} / \mathrm{m} 3$ ) will the ship displace?
2) How much mass can a $1,000,000 \mathrm{~L}$ balloon lift if the inside temperature of the balloon is 80 ${ }^{\circ} \mathrm{C}$ and the outside air temperature is $20^{\circ} \mathrm{C}$ ?
3) How many 10 L helium balloons would it take to lift a man in an armchair ( 75 kg ) if the density of air is $1.2 \mathrm{~g} / \mathrm{L}$ and the density of helium is $0.1786 \mathrm{~g} / \mathrm{L}$. Assume each balloon has a weight of 3 g . (Hint: First calculate how much extra mass a helium balloon can carry)
