Mystery Elements Cards

Instructions: Print, cut apart on lines and shuffle. Use the numbers in the lower right corner of each card to find the "mystery" element name on the answer sheet.

Mystery Element

- First element in the periodic table
- One of two elements that make up water
- Gaseous at room temperature
- Extremely flammable
- Oxidation state (charge) of +1



Mystery Element

- Diatomic gas
- 79% of our atmosphere is this in gaseous form
- Liquid used in cryogenics and to make ice cream
- One of two elements in ammonia
- Integral part of amino acids



Mystery Element

- Often used to inflate balloons
- Gaseous at room temperature
- Temporarily changes one's voice when a small amount is inhaled
- Second lightest element
- A noble gas



Mystery Element

- Major component of air
- Waste product of photosynthesis
- Human life would not exist without it
- Oxidation state (charge) of -2
- One of its states (isotopes) plays a major role in the protective layer around the Earth



Mystery Element

- Lightest metal element
- Chemically very similar to sodium
- Used in batteries
- Reacts easily with water
- Oxidation state (charge) of +1

Mystery Element

- Halogen
- Poisonous and corrosive gas
- Reacts explosively with hydrogen
- Most reactive and electronegative of all elements
- Used in production of Teflon



Mystery Element

- Alkaline earth metal
- Contains four protons
- Used as a hardening agent in metal alloys
- Emeralds and aquamarines contain this element
- Toxic



Mystery Element

- Colorless inert gas
- Part of the noble gas family
- Second lightest noble gas
- Used to make attention-getting signs
- Contains 10 protons



Mystery Element

- Solid at room temperature
- Brown or black non-metal
- Acts as a Lewis-acid
- Oxidation state (charge) of +3 (very electrophilic)
- Used to make insulating fiberglass



Mystery Element

- Alkali metal
- One of two elements present in salt
- Found in baking soda
- Necessary for life to exist
- Oxidation state (charge) of +1



Mystery Element

- Contains six protons
- Diamond and graphite are composed of this
- Present in all organic compounds; the basis of life
- Used in radioactive dating
- Makes steel when combined with iron



Mystery Element

- Contains 16 protons
- Yellow non-metal
- Found in several amino acids; essential for life
- Associated with a rotten egg smell
- Part of a relatively strong acid used as an industrial raw material

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Mystery Element

- Very abundant on Earth
- Alkaline earth metal
- Used to make mild of magnesia and Epson salts
- Used to make metal alloys
- An important human nutrient found in nuts, seeds and whole grains



Mystery Element

- Used to disinfect water and swimming pools
- Diatomic gas
- One of two elements found in common salt
- Extremely poisonous
- Halogen with a -1 oxidation state (charge)

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Mystery Element

- Very resistant to oxidation
- Strong and lightweight metal
- Used to make automobiles and beverage cans
- Was once considered a precious metal due to the rarity of the pure element
- Oxidation state (charge) of +3

Mystery Element

- Contains 18 protons
- Colorless and odorless
- Commonly used in lighting
- Most common noble gas on Earth
- No confirmed chemical compounds contain this element



Mystery Element

- Similar properties to carbon, but less reactive
- Contains 14 protons
- Second most abundant element in Earth's crust
- Quartz and sand contain this element
- Used in semiconductor industry and a major component of glass and cement



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Mystery Element

- Shares many characteristics with sodium
- Second lightest metal
- Alkali metal that reacts violently with water
- Found in high quantities in bananas
- A shortage of this element in the body leads to a fatal condition known as hypokalemia

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Mystery Element

- Very reactive, solid, non-metal
- Oxidation state (charge) of -3 or +5
- Found in all living cells
- Burns spontaneously in air
- Used to produce fertilizer and by cells to transport energy



Mystery Element

- Soft, gray alkaline earth metal
- Essential for life
- Human nutrient found in dairy products
- Fifth most abundant element in the Earth's crust
- Oxidation state (charge) of +2



Mystery Element

- Transition metal
- Contains 21 protons
- Used to make high-intensity lights
- Used to form metal alloys with aluminum
- Found in rare minerals in Scandinavia

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Mystery Element

- Comes from Latin word "ferrum"
- Essential to all organisms; used in the heme complex in humans
- Most abundant element that makes up the Earth
- Most widely-used metal
- Extracted from hematite and magnetite



Mystery Element

- Similar properties to zirconium
- Used to make very strong, light-weight metal alloys
- Used to make medical implants, aircraft and boats
- As strong as steel, but half the weight
- Corrosion resistant



Mystery Element

- Contains 27 protons
- Corrosive-resistant transition metal
- Oxidation state (charge) of +2 or +3
- Used in radiotherapy
- Comes from the German word "kobold," which means evil spirit



Mystery Element

- Transition metal containing 23 protons
- Used to make metal alloys, particularly steel
- Rare and ductile
- Very toxic
- Only found bound to other elements

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Mystery Element

- Used in coin making; and brass and iron plating
- A US monetary coin shares its name with this element that contains it
- Found in rechargeable batteries
- Belongs to the iron group
- The Earth's core is supposedly composed of this element and iron



Mystery Element

- Reacts immediately with oxygen, forming a protective layer
- Steel gray color
- Helps with sugar metabolism in humans
- Used in making metal alloys as well as in dyes and paints



Mystery Element

- Used in the US one cent piece
- Reddish-gold color
- Known as "chalkos" by the Greeks, "aes Cyprium" by the Romans, and simplified to "cuprum"
- Used by civilizations more than 10,000 years ago
- Used to make brass/bronze metal alloys



Mystery Element

- Essential to steel production
- Contained in potassium permanganate, which is used in medicine
- Nutrient in all of life
- Found primarily in pyrolusite

Part of the iron group of elements



Mystery Element

- Contains 30 protons
- Fourth most common metal used on Earth
- Necessary for life; found in animal proteins
- Human deficiency leads to memory and sensory loss as well as stunted brain development
- Oxidation state (charge) of +2

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